



# 2025 ICH Q2(R2)/Q14 Training based on Training Materials

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The goals of this presentation is to help the audience

1 Understand Overall structure of Q2(R2) and Q14

2 Find point/theme of interest

By presenting

Overall structure of Q2 and Q14 guidelines and selected slides from the training materials.

*This presentation contains the presenter's personal opinion and does not represent the formal ICH position nor the formal MHLW's position.*

# Outline of This Presentation

- **Overview of the two guidelines**

- Structure of the training materials and read suggestions

- Key points with examples/cases

Validation study vs. Validation test (Validation strategy, Development data, MODR, Platform)

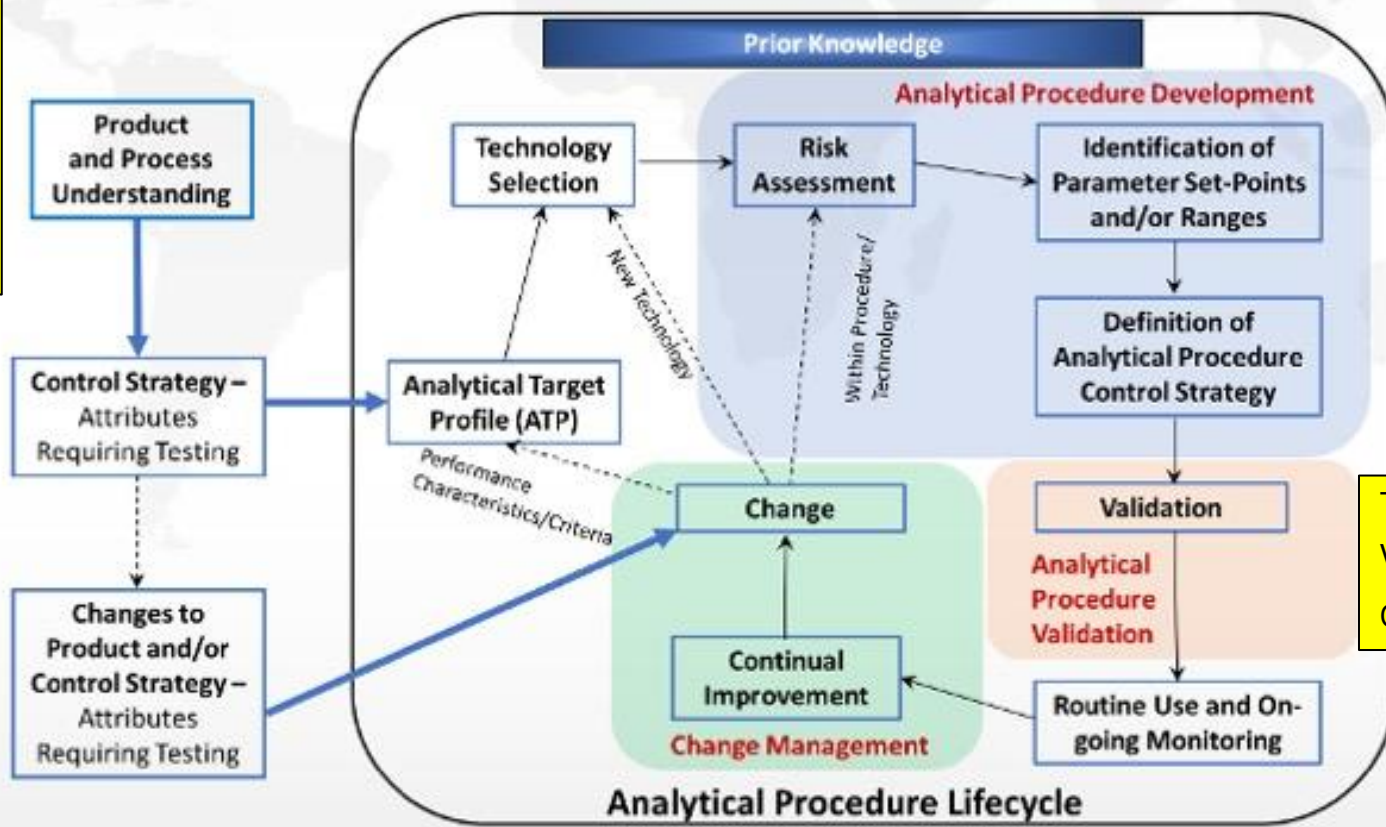
Lifecycle(ATP, Development, Validation, ECs, Changes) example(s)

- Conclusions

## Chapter 2.2: Analytical Procedure Lifecycle

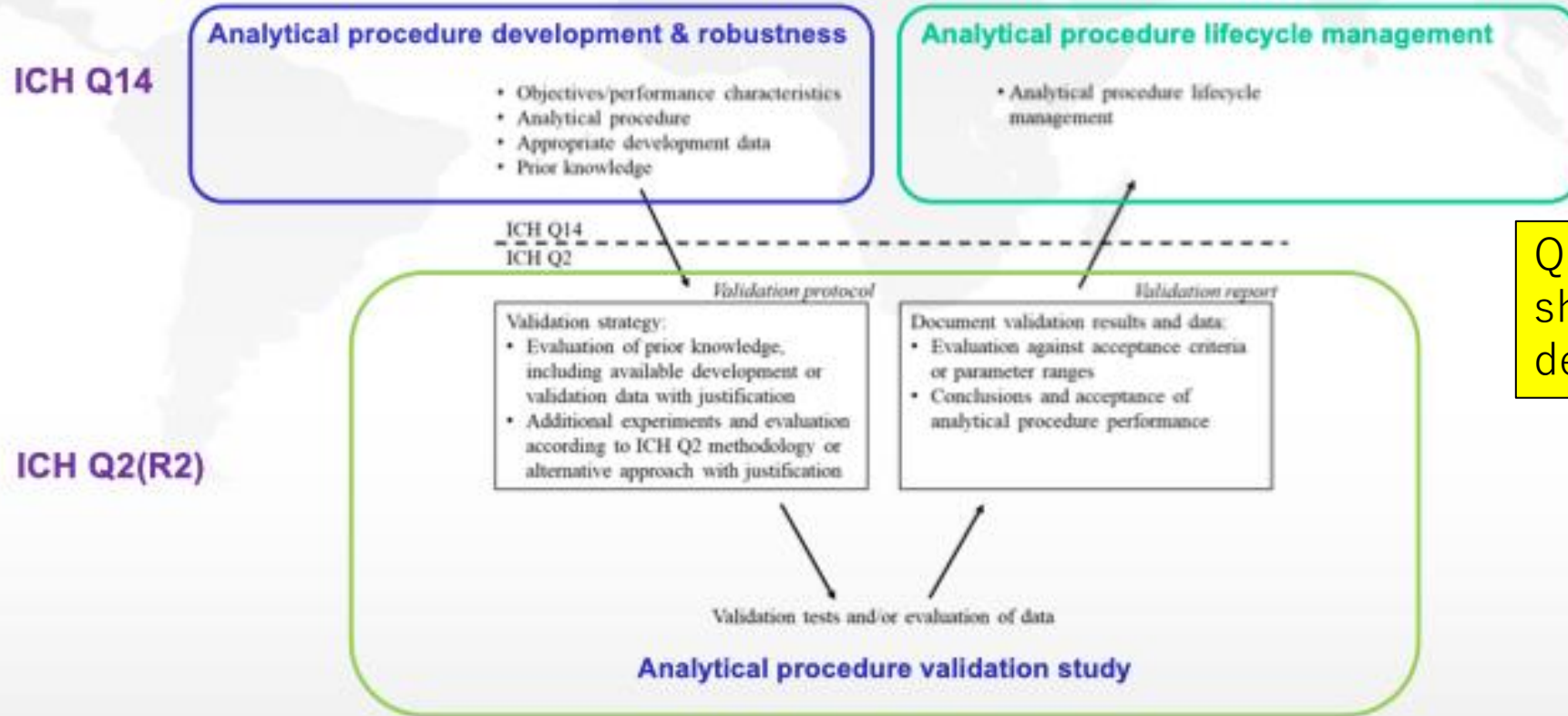
Q14 has this lifecycle figure

Key points in product development shown



Technical details of Validation is described in Q2

- ICH Q2(R2) analytical procedure validation is an element of the analytical procedure lifecycle described in ICH Q14



Q2 has this chart showing linkage between development and validation

Adapted from ICH Q2(R2) Figure 1: Validation study design and evaluation

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  - Lifecycle(ATP, Development, Validation, ECs, Changes)  
example(s)
- Conclusions

## ICH Q2(R2) / Q14 Training Modules Map of Contents

High level of cross linkage

	Module Number	Q2(R2) Q14	Contents	Linkage Cross Reference		
41	Module 1	Q2(R2)/ Q14	Step 4 Presentation for ICH Q2(R2) and ICH Q14			
33	Module 2	Q2(R2)	<b>Fundamental principles of ICH Q2(R2)</b>			
			Part A	Analytical Procedure Validation Strategy	Module 3 Part B Use of development data Module 4 Part B, Part E and Part F Module 7 MODR.	
			Part B	Details of Validation Terms		
			Part C	Combined Accuracy and Precision	Module 3 Part B Use of Confidence Intervals	
			Part D	Considerations when Setting Performance Criteria		
102	Module 3	Q2(R2)	<b>Practical Applications of ICH Q2(R2)</b>			
			Part A	ICH Q2(R2) Annex 1 and 2	Table 3 Separation techniques	
					Table 5 Dissolution with HPLC	Module 5 Part E
					Table 8 Quantitative PCR	
					Table 9 Particle size measurement	
			Part B	Other Validation Topics	Platform Analytical Procedures	Module 7 Platform Analytical Procedures
					Use of Confidence Intervals	
					Use of Replicates	
	Use of Development Data					
			Single Point Calibration			
			Extrapolation of Validation Range			
			Quantitative Test vs. Limit Test	Module 5 Part E		
46	Module 4	Q14	<b>ICH Q14 General Considerations</b>			
			Part A	Minimal and Enhanced Approach		
			Part B	Analytical Procedure Lifecycle		
			Part C	Analytical Target Profile (ATP)		
			Part D	Risk Assessment in Analytical Procedure Development	Module 5 Part C Module 7	
			Part E	Robustness and Parameter Ranges	Module 2 Part A Module 7 MODR.	
			Part F	Analytical Procedure Control Strategy	Module 2 Part A Module 7	

Module Number	Q2(R2)/ Q14	Contents	Linkage Cross Reference	
60	Module 5 Q14	<b>Further Concepts in ICH Q14</b>		
		Part A	Established Conditions and the Link to ICH Q12	Module 7
		Part B	Change Management: Identification of ECs/Reporting Categories and the Use of the Decision Tree	Module 7
		Part C	Knowledge and Risk-based Change Management	Module 4 Part D
		Part D	Explanation of ICH Q14 Tables 1 and 2, Implementation of Changes, and Bridging Studies	
		Part E	Submission Requirements in ICH Q14 Chapter 10	Module 3 Part A (Table 5) Module 6 Module 7 and Module 7 NIR
31	Module 6 Q2(R2)/ Q14	<b>Multivariate Analytical Procedure</b>		
		Introduction		
		Examples	Raman spectroscopy for identity testing	
			NIR for assay	Module 5 Part E Module 7 NIR
	Raman spectroscopy for glucose testing			
88p	Module 7 Q2(R2)/ Q14	<b>Additional Case Studies and Examples</b>		
		Impurity	Case Study - Measurement of Stereoisomers as Specific Process Related Impurities in a Small Molecule Drug Substance	
		Potency	Case Study - Measurement of Potency for an anti-TNF-alpha Monoclonal Antibody	
		MAM	Case Study - MAM by Peptide Mapping LC-MS	
		NIR	Case Study - At-line Assay of Core Tablets by Multivariate Analytical Procedure in Continuous Manufacturing of a Drug Product	Module 5 Part E Module 6
		Platform Analytical Procedures	Example - Platform Analytical Procedure, Determination of High Molecular Weight Species in Monoclonal Antibody Products	Module 3 Part B Platform Analytical Procedures
		MODR	Example - Application of the Enhanced Approach Using DoE Studies - Establishment and Validation Options for MODRs	Module 2 Part A Module 4 Part E

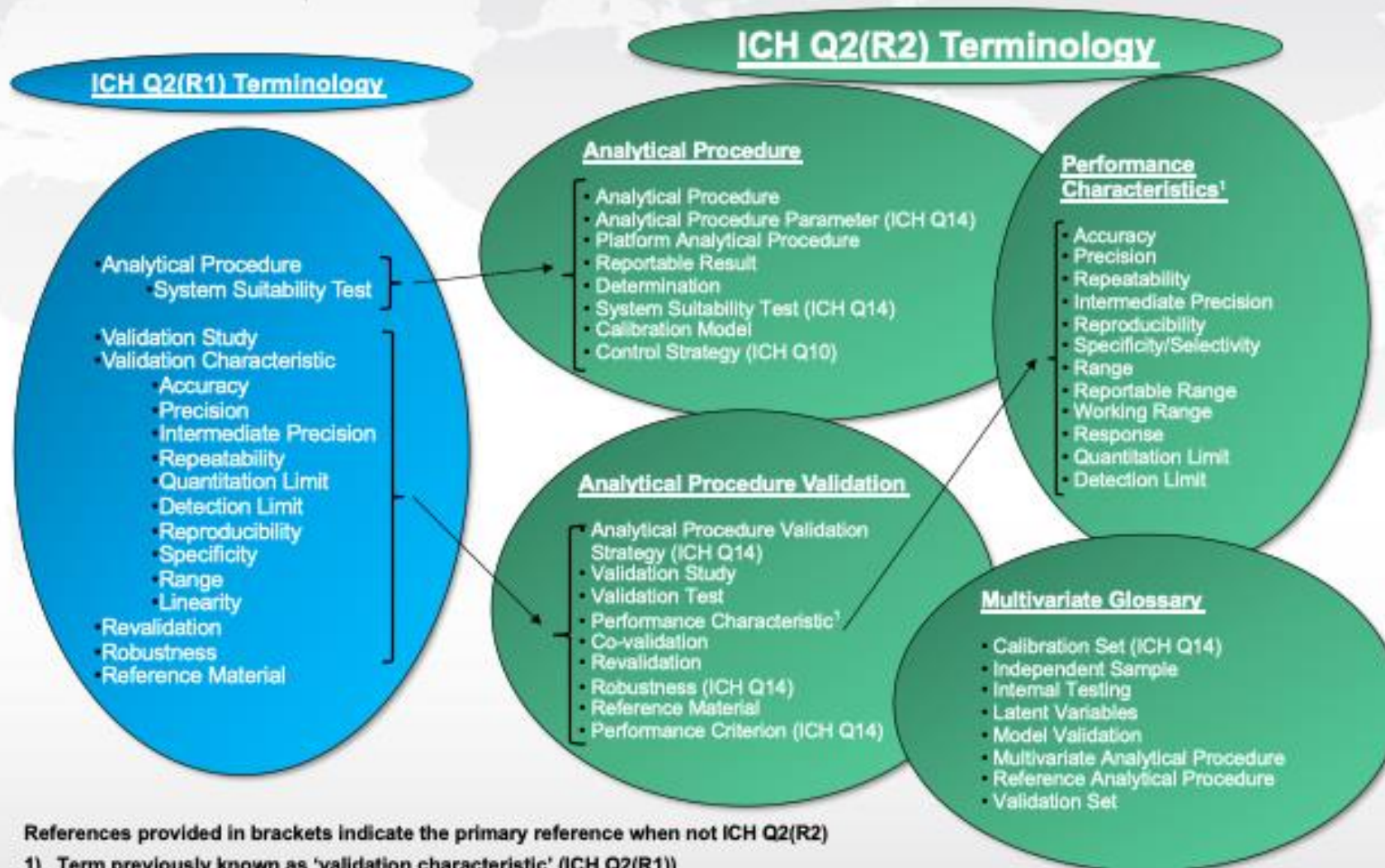
NIR: Near-infrared spectroscopy  
MAM: Multi-attribute Methods  
MODR: Method Operable Design Region

# Read Suggestions

- If you need only a very high level understanding on Q2 and Q14, look at training module 1.
- If you like to get comprehensive detail understanding, pick one of four case studies in module 7 and read corresponding descriptions module 2 through 6.
- Terms used in the guidelines are mostly defined. For clear understanding of those, check definitions first. Also, interrelations between those terms are important in order to understand the contents, the following figure and chart may be useful.

# ICH Q2(R2) / Q14 Training Module 2

## Overview of ICH Q2 Terminology



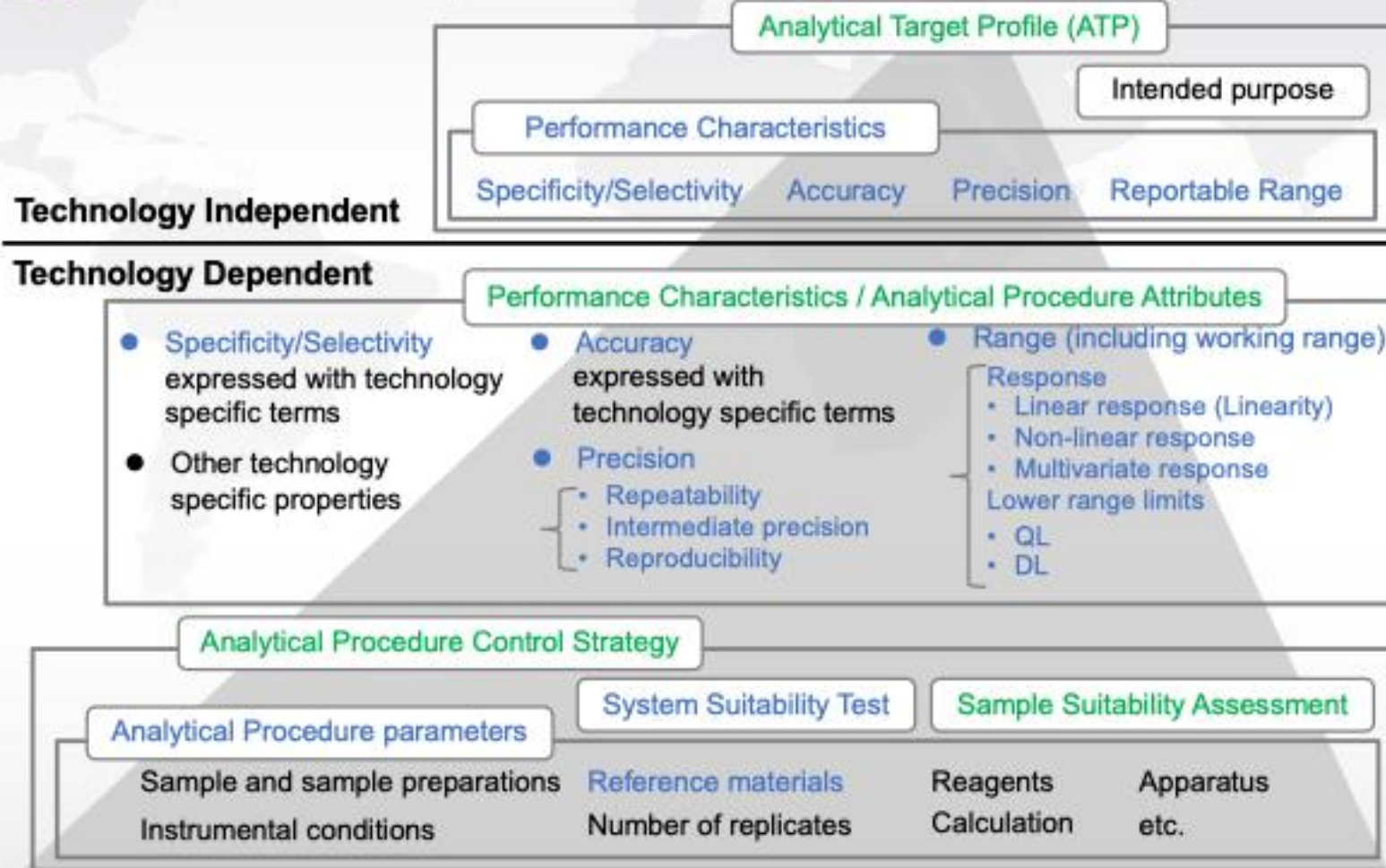
References provided in brackets indicate the primary reference when not ICH Q2(R2)

1) Term previously known as 'validation characteristic' (ICH Q2(R1))

# Terminology Hierarchy

## Note

This figure illustrates terminology hierarchy by merging both ICH Q2(R2) and ICHQ14 terminologies from the view of using an ATP, an element of the enhanced approach as described in ICH Q14.



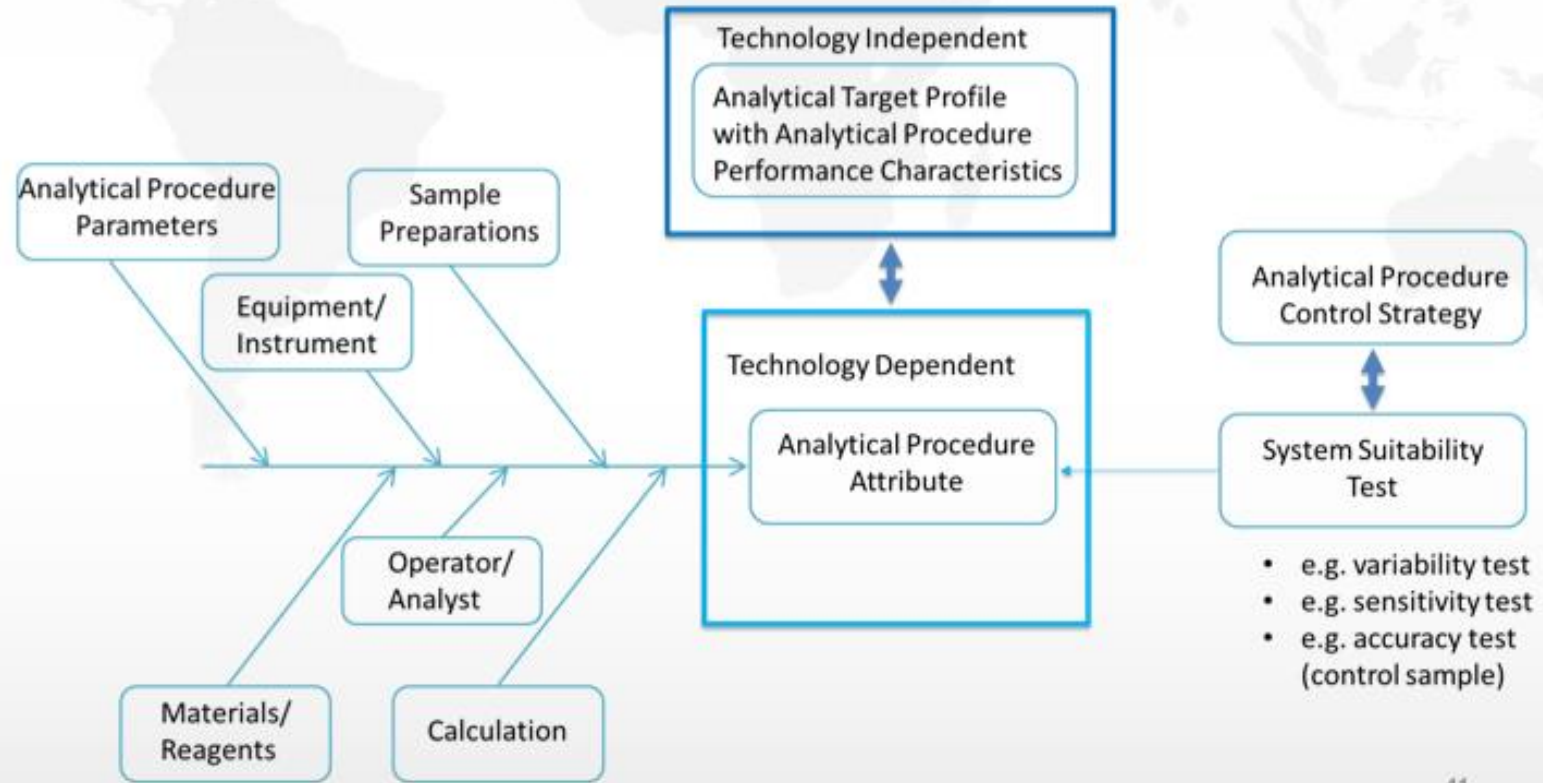
Output related

Input related

Terms defined in ICH Q2(R2) are colored in blue, and those defined only in ICH Q14 are colored in green.

# Analytical Procedure Control Strategy

## Risk Assessment, ATP and Analytical Procedure Control Strategy



## Analytical Procedure Development Flow in Line with ICH Q14

### Note

When using an ATP a "Performance characteristic" is a technology-independent description of characteristic with an associated and defined acceptance criteria. Once a technology is selected, technology-dependent performance characteristics can be determined, which are defined as "Analytical Procedure Attributes" in ICH Q14.

### Analytical Target Profile (ATP)

Consisting of below elements

- Intended purpose of analytical procedure
- Details on quality attribute to be tested
- **Performance characteristics** and associated **performance criteria**

### Technology selection

### Analytical procedure development

Risk assessment: Identifying **analytical procedure parameters** with potential impact on performance, assessing the potential impact, and identifying **analytical procedure parameters** to be investigated experimentally.

Robustness evaluation: Testing by deliberate variations of **analytical procedure parameters** considering duration of analysis.

Analytical procedure parameter ranges: Investigating the impact of **analytical procedure parameter** (input) ranges to **analytical procedure attributes** (output) and associated criteria that can be derived from an ATP.

Analytical procedure control strategy: Includes **analytical procedure parameters** needing control and **SST**. SST is designed to verify selected **analytical procedure attributes**.

### Validation study

Validation protocol: A written plan describing the analytical procedure to be validated, **performance characteristics / analytical procedure attributes** and associated criteria derived from an ATP, **validation tests** to be conducted, participating sites etc. Validation protocol is designed based on or includes **analytical procedure validation strategy** considering prior knowledge and existing data.

Validation tests and/or evaluation of data

Validation report: Document of validation results and data; and conclude suitability

Terms defined in ICH Q2(R2) are colored in **blue**, and those defined only in ICH Q14 are colored in **green**. <sup>17</sup>

**VALIDATION STUDY**  
An evaluation of **prior knowledge, data, deliberate experiments**

**VALIDATION TEST**  
Validation tests are **deliberate experiments**

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**Validation study vs. Validation test (*Validation strategy, Development data, MODR, Platform*)**

Lifecycle(ATP, Development, Validation, ECs, Changes) example(s)

- Conclusions

**Documented study designed to provide sufficient evidence that the analytical procedure meets its objectives**

- **Protocol:**

- Intended purpose of the analytical procedure
- Based on the intended purpose, provide the appropriate *Performance Characteristics* to be validated (as per ICH Q2(R2) Table 1) and the associated *Performance Criteria*
- Overview of analytical procedure validation strategy
  - Justification of appropriateness of any prior knowledge
    - "Suitable data derived from development studies can be used as part of validation data" (ICH Q2(R2))
    - "In cases where prior knowledge is used (e.g., from development or from previous studies), appropriate justification should be provided" (ICH Q2(R2))
    - See also Module 3, Part B (Use of Development Data)
  - Experimental design to assess performance characteristics for which suitable prior knowledge is not sufficient or is not available.
    - Experimental design should reflect the number of replicates used in routine analysis to generate a reportable result. If justified, it may be acceptable to perform some validation tests using a different number of replicates or to adjust the number of replicates in the analytical procedure based on data generated during validation.

- **Report**

- Results of the study, including comparison to Performance Criteria
- A tabular validation summary to demonstrate ICH Q2(R2) compliance may be useful
- A conclusion regarding the suitability of the procedure for its intended use should be included

### VALIDATION STUDY

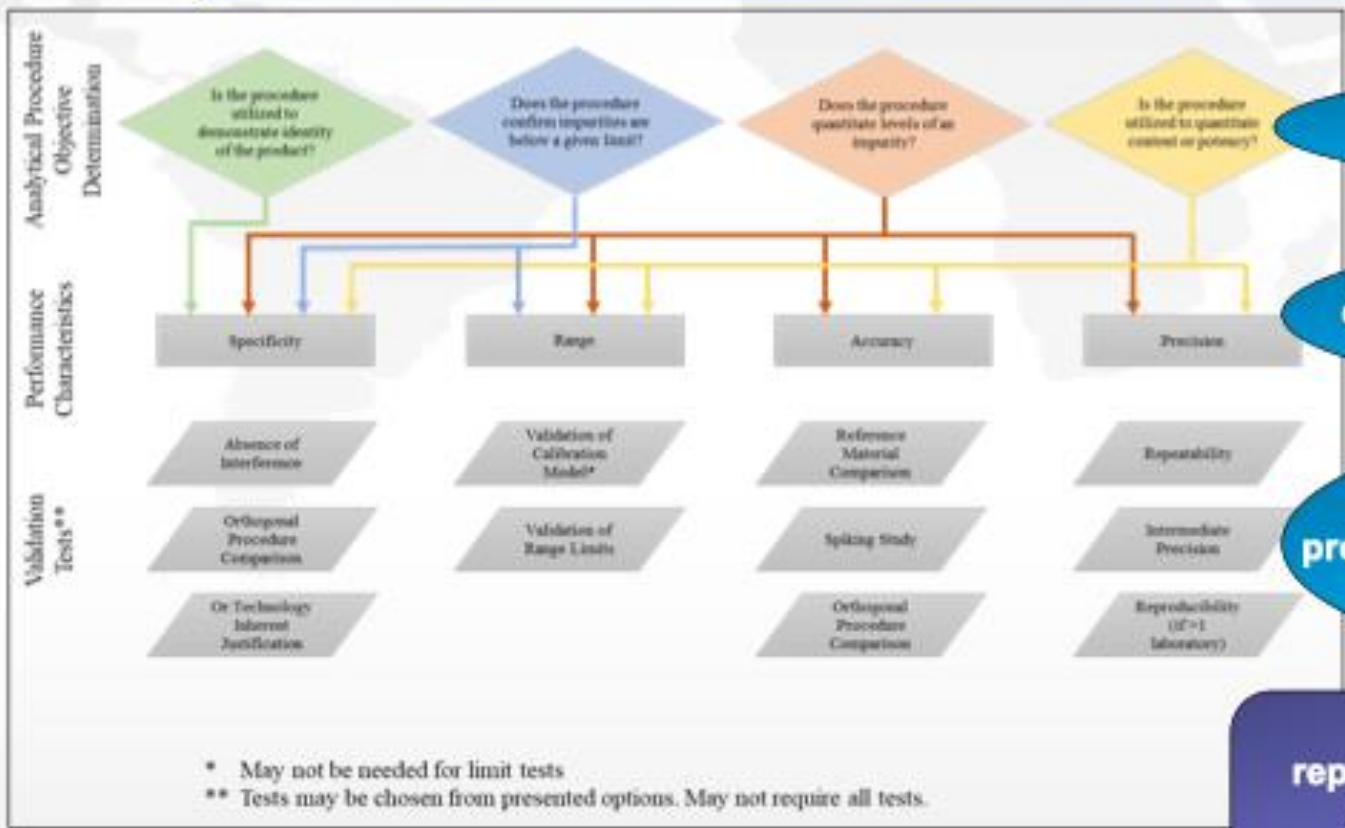
An evaluation of  
prior knowledge, data ,  
deliberate experiments

### VALIDATION TEST

Validation tests are  
deliberate experiments

# ICH Q2(R2) Framework

**ICH Q2(R2) provides a framework for the approach to analytical procedure validation, which can be applied irrespective of the measured quality attribute or the technology used.**



\* May not be needed for limit tests  
\*\* Tests may be chosen from presented options. May not require all tests.

**Objectives of the analytical procedure are determined**

**Relevant performance characteristics are selected based upon the intended use of the analytical procedure**

**Suitable validation test(s) are chosen based on specific procedure and product considerations, e.g., available reference materials, inherent properties of the technology used.**

**Figure 2 (Annex 1) provides a flow chart representation of the performance characteristic selection from Table 1, as well as example validation tests that may be considered for each characteristic.**

ICH Q2(R2) Figure 2: Examples of relevant validation tests based on the objective of the analytical procedure

This Annex gives different **evaluation** types for **Performance Characteristics**

## Annex 2 - Illustrative Examples for Analytical Techniques

### Specific non-binding examples for common techniques :

- Separation techniques (*e.g.*, HPLC, GC, CE) for impurities or assay
- Separation techniques with relative area quantitation, (*e.g.*, product-related substances such as charge variants)
- Elemental Impurities by ICP-OES or ICP-MS
- Dissolution with HPLC as product performance test for an immediate release dosage form
- Quantitative <sup>1</sup>H-NMR (internal standard method) for the assay of a drug substance
- Binding assay (*e.g.*, ELISA, SPR) or cell-based assay for determination of potency relative to a reference
- Quantitative PCR (quantitative analysis of impurities in drug substances or products)
- Particle size measurement (dynamic light scattering; laser diffraction measurement) as a property test
- NIR analytical procedure for core tablet assay
- Quantitative LC/MS analysis of trace impurities in product

The dissolution example uses development data for specificity

The qNMR example employs calibration for precision

## ICH Q2(R2) Annex 2 Examples

- The tables presented in ICH Q2(R2) Annex 2 provide example approaches for analytical procedure validations.
  - The technologies and approaches presented were constructed to illustrate potential applications of the principles contained within the guideline and are not exhaustive.
  - The examples in Annex 2 are not intended to be mandatory, and alternative approaches (fulfilling the intent of the guideline) may also be acceptable.
- Examples have been elucidated for four of the technologies contained in the tables in Annex 2.
  - These examples provide an additional layer of information beyond that in Annex 2, and exemplify the data which may be collected during analytical procedure validation.
  - This additional information is not intended to be mandatory, and alternative approaches (fulfilling the intent of the guideline) may also be acceptable.
- The following slides present example validation data relating to:
  - ICH Q2(R2) Annex 2, Table 3:
    - Separation techniques with relative area quantitation (e.g., product-related substances such as charge variants).
  - ICH Q2(R2) Annex 2, Table 5:
    - Dissolution with high-performance liquid chromatography (HPLC) as product performance test for an immediate release dosage form.
  - ICH Q2(R2) Annex 2, Table 8:
    - Quantitative polymerase chain reaction (qPCR) (quantitative analysis of impurities in drug substances or products).
  - ICH Q2(R2) Annex 2, Table 9:
    - Particle size measurement (dynamic light scattering; laser diffraction measurement) as a property test.

# ICH Q2(R2) / Q14 Training Module 3

## Example Validation Data for Separation Techniques with Relative Area Quantitation (Annex 2, Table 3)

Technique	Separation techniques with relative area quantitation, (e.g., product-related substances such as charge variants)
Performance characteristic	Validation study methodology
Specificity / Selectivity	<p><u>Absence of relevant interference:</u> With product, buffer, or appropriate matrix, and between individual peaks of interest</p> <p>Demonstration of stability-indicating properties through appropriate forced degradation samples if necessary</p>
Precision	<p><u>Repeatability:</u> Replicate measurements with 3 times 3 levels across the reportable range or 6 times at 100% level, considering peak(s) of interest</p> <p><u>Intermediate precision:</u> e.g., different days, environmental conditions, analysts, equipment</p>
Accuracy	<p>Comparison with an orthogonal procedure and/or suitably characterised material (e.g., reference material)</p> <p>or</p> <p>Accuracy can be inferred once precision, linearity and specificity have been established.</p> <p>or</p> <p>Spiking studies with forced degradation samples and/or suitably characterised material</p>
Reportable Range	<p><u>Validation of calibration model across the range:</u></p> <p><u>Linearity:</u> Between measured (observed) relative result versus theoretically expected relative result across specification range(s), e.g., by spiking or degrading material</p> <p><u>Validation of lower range limits:</u> QL (and DL) through a selected methodology (e.g., signal-to-noise determination)</p>
Robustness and other considerations (performed as part of analytical procedure development as per ICH Q14)	<p><u>Deliberate variation of relevant parameters, e.g.,</u></p> <p>Sample preparation: extraction volume, extraction time, temperature</p> <p>Separation parameters: column/capillary lot, mobile phase/buffer composition and pH, column/capillary temperature, flow rate, detection wavelength</p> <p>Stability of sample and reference material preparations</p> <p><b>Relative Response Factors</b></p> <p>If the analyte has a different response from the reference material (e.g., a different specific UV absorbance), relative response factors should be calculated using the appropriate ratio of responses. This evaluation may be performed during validation or development, and should use the finalised analytical procedure conditions and be appropriately documented</p> <p>If the relative response factor is outside the range 0.8-1.2, then a correction factor should be applied. If an impurity/degradation product is overestimated, it may be acceptable not to use a correction factor</p>

The bottom box is included in Q2 after debate by the E W G

## Determination of Monoclonal Antibody Charge Variants by Ion Exchange Chromatography

### Liquid Chromatography (LC) procedure

- Column: weak cation-exchange resin, 250 mm × 4.0 mm (10 µm)
- Gradient elution: mobile phase A (phosphate buffer),  
mobile phase B (phosphate buffer, sodium chloride)
- Sample concentration: 1 mg/mL

Parameter	Set point
Flow rate	1.0 mL/min
Column temperature	40°C
Detection	Ultraviolet (UV) at 280 nm
Run time	80 min
Injection volume	50 µL

**Acceptance criteria defined for % acidic peaks, % main peak, % basic peaks.**

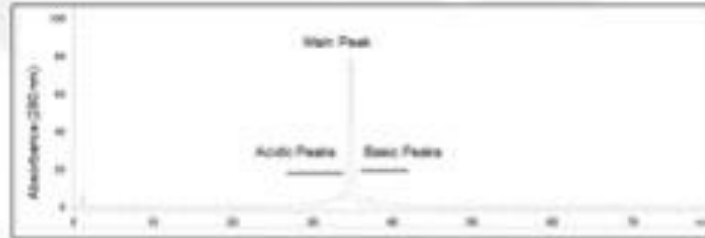
Other performance Characteristics are shown in the module. But not shown in this presentation

### Sample matrix interference

Analyse the separation and matrix component interference of the drug substance (DS).

- DS shows clear separation order of acidic peaks, main peak and basic peaks.
- No significant interference from sample matrix components in the chromatographic region of interest.

Chromatogram of the DS



Chromatogram for DS formulation buffer



### Stability-indicating properties

Comparison of chromatograms obtained with reference material and stressed sample.

- The chromatogram of the reference material should be distinguishable from that of stressed sample by visual comparison.

Relative peak area (%)

	Acidic peaks		Main peak	Basic peaks		
	Peak 1	Peak 2	Peak 3	Peak 4	Peak 5	Peak 6
Reference material	2.23	9.88	12.00	65.17	8.85	1.86
Stressed sample	3.55	11.00	11.66	46.82	11.54	3.35

Additional new peaks\*: Relative peak area (%)

	Peak 7	Peak 8	Peak 9	Peak 10	Peak 11	Sum
Stressed sample	1.78	1.03	3.56	2.63	2.98	12.08

\*Table contains extra peaks labelled independently on whether the new peaks are acidic or basic peaks

The ICH Q2(R2) defines the following:

**From Chapter 2 “General Considerations for Analytical Procedure Validation”:** “Suitable data derived from development studies (see ICH Q14) can be used as part of validation data”.

**From Chapter 2.1 “Analytical Procedure Validation Study”:** “In cases where prior knowledge is used (e.g., from development or from previous studies), appropriate justification should be provided”.

Prior knowledge and/or data generated during development, e.g., while establishing proven acceptable ranges (PAR) or method operable design regions (MODR), can be used as part of the validation study.

MODR is explained in modules 2, 4 and

The broader the prior knowledge and the knowledge about the relationship between analytical parameters and analytical performance characteristics, the higher the chances to be able to use these data instead of generating them again during the validation study.

## Use of Development Data as part of Validation Data

Some examples on usage of development data are following:

- **Selectivity and Specificity:** Using specificity data generated during analytical procedure development as validation evidence.
- **Linearity:** Incorporation of linearity data from the development phase to demonstrate the range of the analytical procedure during validation.
- **Lower Range Limit (DL/QL):** Adapting sensitivity evaluations from development for validation, showing the ability of the analytical procedure to detect and quantify low amounts of analytes.
- **Robustness:** Applying robustness testing data from the analytical procedure development stage to demonstrate analytical procedure reliability under varied conditions.
- **Relative Response Factors:** This evaluation may be performed during validation or development, should use the finalised analytical procedure conditions and should be appropriately documented.
- **System Suitability Testing (SST):** Using SST data from development experiments to confirm system performance during analytical runs.
- **Sample Stability:** Utilising stability data from development studies to demonstrate stability during sample handling and execution of the analytical procedure.

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**Lifecycle**(ATP, Development, Validation, ECs, Changes) example(s)

- Conclusions

# Module 7 shows the following Four case studies

**1 Impurity;** MEASUREMENT OF STEREOMERS AS SPECIFIC PROCESS RELATED IMPURITIES IN A SMALL MOLECULE DRUG SUBSTANCE

**2 Potency;** MEASUREMENT OF POTENCY FOR AN ANTI-TNF-ALPHA MONOCLONAL ANTIBODY

**3 MAM;**MAM(Multi-attribute Method) BY PEPTIDE MAPPING LC-MS

**4 NIR;** AT-LINE ASSAY OF CORE TABLETS BY MULTIVARIATE ANALYTICAL PROCEDURE IN CONTINUOUS MANUFACTURING OF A DRUG PRODUCT

ATP, Technology Selection, Analytical Procedure Development, Analytical Procedure, Analytical Procedure Validation,

Description of ECs/Reporting categories/Justification,

and Change Management/Bridging Study Change Management/Examples

## Analytical Target Profile (ATP)

- A prospective summary of the performance characteristics describing the intended purpose and the anticipated performance criteria of an analytical measurement.
- The intended purpose should be sufficiently descriptive for the measurement of the attribute and the use of that result.
- Facilitates the selection of the technology, the procedure design and development as well as the subsequent performance monitoring and continual improvement of the analytical procedure.
- Multiple available analytical techniques may meet the performance requirements.
- Maintained over the lifecycle and can be used as basis for lifecycle management.
  - An ATP could also be retrospectively established.

**Table 1: Analytical Target Profile**

<b>Intended Purpose</b>		
Quantitation of the six stereoisomers A - F in Sakuratinib Maleate DS for release testing		
<b>Link to critical quality attribute (CQA) (Stereoisomeric Purity)</b>		
The analytical procedure should allow for the quantitation of the individual stereoisomers A - F and determination of the total sum to verify the CQA Stereoisomeric Purity $\geq 99.0\%$		
<b>Characteristics of the Reportable Results</b>		
<b>Performance Characteristics</b>	<b>Acceptance Criteria*</b>	<b>Rationale</b>
Accuracy	80 - 120% average recovery of spiked DS with Impurities A - E (specified at NMT 0.1% each) 90 - 110% average recovery of spiked DS with Impurity F (specified at NMT 0.5%)	For example, at a specification level of 0.1%, 20% bias would lead to a variation of the analytical result of 0.02%, which was found acceptable for a release decision.
Precision	Intermediate Precision relative standard deviation (RSD): Impurities A - E $\leq 15\%$ Impurity F $\leq 10\%$	In a similar fashion, values for precision were derived. The recovery criteria for accuracy were set with respect to the reported result and taking into consideration any correction or response factors
Specificity	Analytical procedure should be able to quantitate impurities A - F in presence of other likely process related substances or DS degradation products with an acceptable bias of not more than 0.02%	Potential interference with quantitation of specified impurities by other regular components in the sample
Reportable Range	Impurities A - E: at least 0.05 - 0.12% Impurity F: at least 0.05 - 0.6%	Reporting threshold to 120% of specification limit

ATP example;  
Extracted from module 7 IMPURITY

Link to CQA

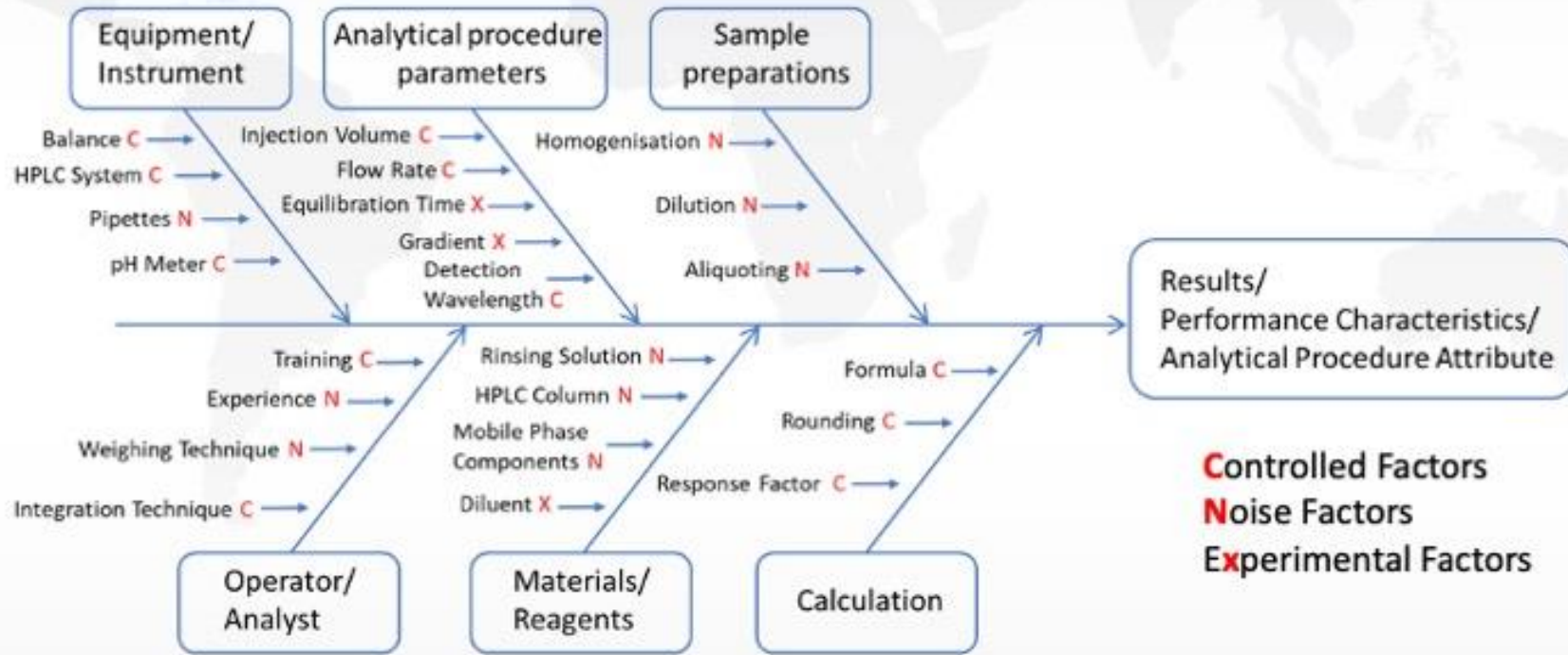
Technology is NOT selected yet

For necessary performance characteristics, acceptance criteria are determined

# Robustness and Parameter Ranges

This Ishikawa diagram is from module 4.

Case studies in module 7 have specific Ishikawa diagram



**Risk assessment can inform the selection of parameters to investigate during the robustness study.**

## Risk Reduction for Changes (ICH Q14 Table 1)

ICH Q14 Table 1 illustrates the relationship between risk and prior knowledge when designing studies in support of a proposed change.

Lower risk enables confirmatory studies to support the change.

Increased risk drives the need for more in-depth studies to support the change.

Prior knowledge can be utilised to inform study design.

Reduced prior knowledge requires comprehensive study.

Table 1: Relationship between knowledge (understanding), risk and extent of studies for changes to analytical procedures

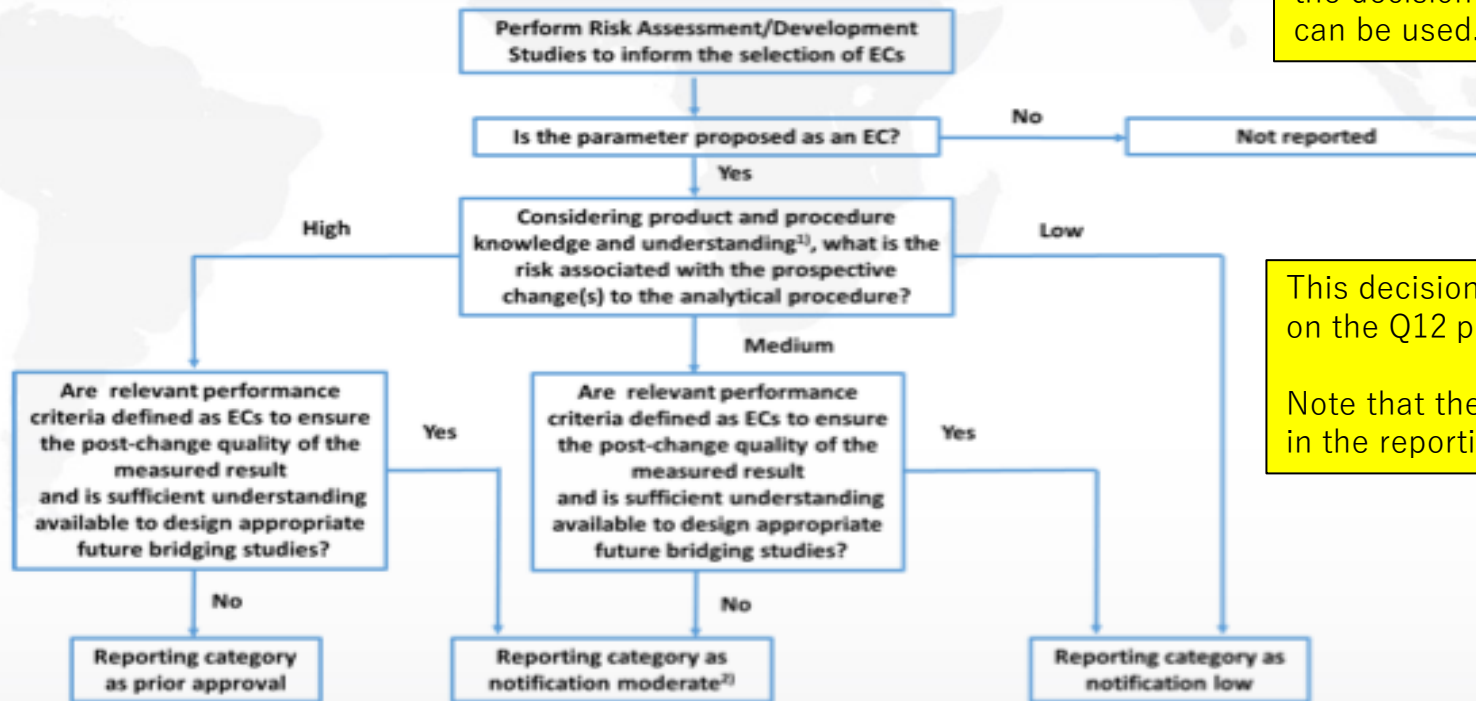
		Risk associated with the change	
		Low	High
Knowledge <sup>1)</sup>	High	Prior knowledge or confirmatory study according to a study plan derived from prior knowledge	In depth study according to a study plan derived from prior knowledge
	Low	Confirmatory study according to a study plan	In depth study according to a study plan

1) As described in ICH Q10

## Analytical Procedure Control Strategy: Established Conditions (ECs) for Analytical Procedures

- In line with ICH Q12, ECs are legally binding information considered necessary to assure product quality.
  - Any change to ECs necessitates a submission to the regulatory authority.
  - ECs are proposed and justified by the applicant and approved by the regulatory authority.
  - They can be identified using tools in Chapter 2 including risk assessment, prior knowledge, and learnings from uni- and multi-variate experimentation.
- The nature and extent of ECs depends on the development approach, complexity of the analytical procedure and demonstrated understanding.

# Lifecycle Management and Post-Approval Changes of Analytical Procedures (ICH Q14 Figure 2)



When ECs are identified with Enhanced Knowledge, the decision tree in this chart can be used.

This decision tree is written based on the Q12 principle.

Note that there are regional differences in the reporting categories

1) Including analytical procedure control strategy.

2) In some cases, moderate risk changes proposed by the company may require prior approval based on health authority feedback.

**Post-approval Change**  
Technique from chiral HPLC to chiral SFC

**Reassessment of the risk of the change with the following points considered**

- **Relevance of the test:** High, control of the CQA (quantitation of stereoisomers)
- **Complexity of the test:** well-established technology
- **Extent of the change:** related technologies; prior knowledge of chemistry, process and impurities informs the procedure development

Estimated risk: High

**Development approach for chiral SFC procedure**

- Enhanced development approach of the same principles as completed for the HPLC procedure was followed
- Risk assessment was conducted to identify the analytical procedure parameters that can impact performance of the procedure
- Modelling and/or multi-variate experiments were performed to explore ranges and interactions between identified analytical procedure parameters
- An analytical procedure control strategy was defined

**Re-confirmation of the following points**

- **Adherence to criteria for relevant performance characteristics:** Those are defined as ECs
- **Sufficient information or prior knowledge to design appropriate bridging studies:** Yes

Risk assessment result before bridging study execution  
is in agreement with the submitted risk category.  
Overall risk category: Medium

**Execution of analytical procedure development**

- Procedure parameters for chiral SFC were determined through knowledge of the process and impurities, and enhanced practices for development
- Analytical procedure description (including AP control strategy (SST)) was finalized

**Demonstration of analytical procedure performance**

- The procedure was validated by a technology specific validation protocol

Risk assessment result before bridging study execution  
is in agreement with the submitted risk category.  
Overall risk category: Medium

**Execution of analytical procedure development**

- Procedure parameters for chiral SFC were determined through knowledge of the process and impurities, and enhanced practices for development
- Analytical procedure description (including AP control strategy (SST)) was finalized

**Demonstration of analytical procedure performance**

- The procedure was validated by a technology specific validation protocol
- Through above studies, relevant performance characteristics and the analytical procedure attributes were evaluated and confirmed
- Execution of bridging study (ICH Q14 Table 2)

**Conclusions**  
The impact of the change was determined based on the bridging study result

- **Impact on the test performance:** The relevant performance characteristics and the analytical procedure attributes met their criteria
- Bridging study results met all acceptance criteria

**Regulatory reporting**  
Reported according to the pre-agreed reporting category  
in registration as notification moderate and submitted  
suitable documents.

From module 7, impurity.  
Change #2; HPLC to SFC

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Lifecycle(ATP, Development, Validation, ECs, Changes) example(s)

- **Conclusions**

## Considerations

- The ICH Q14 and ICH Q2(R2) guidelines should be applied in conjunction with other existing and prospective ICH “Q” guidelines, including ICH Q8-Q13.
- Analytical procedure development can be performed following a minimal or enhanced approach. Though not mandatory, the use of individual elements of the enhanced approach is encouraged to be applied in an as needed basis.
- Tools and enablers discussed in ICH Q12 are applicable to analytical procedures, irrespective of the development approach.
- Examples in ICH Q2 Annex 2 describe common analytical technologies. The principles, however, can be applied in a similar fashion to other analytical technologies.

Use Q2 and Q14 together.  
And with others, Q8-Q13.

The enhanced approach  
yields  
comprehensive knowledge

## Conclusions

- The ICH Q14 and ICH Q2(R2) guidelines establish harmonised scientific and technical principles for analytical procedures over the entire analytical procedure lifecycle.
- Applying principles described in ICH Q14 can improve regulatory communication between industry and regulators and facilitate more efficient, sound scientific and risk-based approval as well as post-approval change management of analytical procedures.
- ICH Q2(R2) will continue to provide a general framework for the principles of analytical procedure validation and has been modernised to include newer technologies (e.g., for biological products or multivariate analytical procedures).

Thank you  
for the attention!



Q2/Q14 EWG Group Photo in Vancouver, June 2023